

14338

California Public Utilities Commission

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web explain how a molecule that contains polar bonds can be nonpolar answer problem 6 2 2 6 2 2 which of the following molecules and ions contain polar bonds which of these molecules and ions have dipole moments a clf 5 b clo 2 clo 2 c tecl₂ 4 tecl₄ 2 d pcl₃ 3 e sef₄ 4 f ph₂ 2 ph₂ g xef₂ answer problem 6 2 3 6 2 3

7 6 molecular structure and polarity chemistry

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web the relative electronegativities of the bonded atoms is h c cl and so the bond moments all point toward the cl end of the molecule and sum

to yield a considerable dipole moment the molecules are relatively polar for molecules of high symmetry such as bf₃ trigonal planar ch₄ tetrahedral pf₅ trigonal bipyramidal and sf₆ 6 1 electronegativity and polarity problems - Aug 15 2023

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chemical bonds may be classified as being either polar or nonpolar the difference is how the electrons in the bond are arranged key takeaways what is a polar bond in chemistry a polar bond is a type of covalent bond in which the electrons forming the bond are unequally distributed

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