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Jianjun Gao

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web in this article we will discuss polyatomic ions the prefix poly means many so a polyatomic ion is an ion that contains

more than one atom this differentiates polyatomic ions from monatomic ions which contain only one atom examples of monatomic ions include na fe 3 cl and many many others
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web because these ions contain more than one atom they are called polyatomic ions polyatomic ions have characteristic formulas names and charges that should be memorized for example the nitrate ion has one nitrogen atom and three oxygen atoms with an overall charge of 1
[5 3 naming compounds that](#)

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these are called polyatomic ions table 2 lists the formulas charges and names of some common polyatomic ions only one of them the ammonium ion is a cation the rest are anions most of them also contain oxygen atoms so sometimes they are referred to as oxyanions
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polyatomic practice 1 name or write the formula for the following
polyatomic ions
sulfate nitrite
perphosphate
hypoiodite chlorite
phosphate

percarbonate
bromate hyposulfite
permanganate
carbonite co3 2
mno3 1 so5 2 bro2
1 co4 2 po5 3 clo1
io2 1 po4 3 no2 1
so4 2 2

nomenclature for polyatomic ions

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are charged groups
of atoms an
example is
ammonium ion nh4
it has five atoms
one nitrogen and
four hydrogens that
share a charge of 1
the polyatomic ions
remain intact and
parentheses may be
required when using
subscripts for
example
ammonium chloride
is nh4cl and
ammonium sulfide
is nh4 2s
3 4 *polyatomic ions
and formulae for
ionic compounds* -

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web polyatomic ions
have defined
formulas names and
charges that cannot
be modified in any
way table 3 4 1 3 4
1 lists the ion
names and ion
formulas of the
most common
polyatomic ions for
example no 3 no 3
is the nitrate ion it
has one nitrogen
atom and three
oxygen atoms and
an overall 1 charge
**ionic compounds
containing
polyatomic ions
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naming ionic
compounds
containing
polyatomic ions
polyatomic ions are
ions which consist
of more than one
atom for example
nitrate ion no 3
contains one

nitrogen atom and
three oxygen atoms
the atoms in a
polyatomic ion are
usually covalently
bonded to one
another and
therefore stay
together as a single
charged unit
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ion composed of
more than one atom
the ammonium ion
consists of one
nitrogen atom and
four hydrogen
atoms together they
comprise a single
ion with a 1 charge
and a formula of ce
nh 4 the carbonate
ion consists of one
carbon atom and
three oxygen atoms
and carries an
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web ions made up of more than one atom are known as polyatomic ions ions with positive charge are called cations ions with negative charge are called anions list of monatomic ions the ions made of a single atom are called simple ions or monatomic ions frequently asked questions on polyatomic ions list q1 what are 3 examples of polyatomic ions **polyatomic ions flashcards quizlet** - May 24 2022
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them 2 elements in the same family make similar ions **4 9 polyatomic ions chemistry libretexts** - Sep 08 2023
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way as other ionic compounds just put the name of the cation and the name of the anion together if there is more than one polyatomic ion don't add a number to the name *polyatomic ions chemistry socratic* - Feb 01 2023
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an electrostatic potential map of the nitrate ion NO_3^- shows three areas coloured translucent red around the outside of the red oxygen atoms themselves signify the regions of most negative electrostatic potential a polyatomic ion also known as a molecular ion is a covalent bonded set of two or more atoms or of a metal *polyatomic ion definition and examples thoughtco* - Jun 24 2022 web aug 2 2022 polyatomic ion definition a polyatomic ion is an ion composed of two or more atoms a polyatomic ion has either a positive charge cation or negative charge anion examples the hydroxide cation OH^+

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consist of groups of
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because these ions
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ions the structures
names and formulas
of some polyatomic
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and table
polyatomic ions

ionic compounds
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