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1s22s22p63s23p64s23d104p65s24d6 ruthenium 23 kr 5s24d10  
cadmium 24 xe 6s24f145d106p2 lead 25 rn 7s25f146d4  
seaborgium determine if the following electron  
configurations are correct 26

1s22s22p63s23p64s24d104p65s1 no it should be 3d10

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of these isotopes write the complete electron  
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