

# Periodic

DK

**Periodic Table, The: Past, Present, And Future** Geoffrey Rayner-canham,2020-08-04 'This is an an absolutely wonderful book that is full of gems about the elements and the periodic table ... All in all, the book is highly recommended to philosophers of chemistry. As philosophers we have a natural tendency to concentrate on generalities and not to get too involved in the specifics and the details. Above all else, this new book reminds us that such an approach needs to be tempered by a detailed knowledge of the exceptions and features that go against the simplified generalities which we so cherish.' [Read Full Review]Eric ScerriFoundations of Chemistry'Many questions are dealt with in a clearly written way in this stimulating and innovative book. The reader will quickly become interested in the subject and will be taken on tour through this Periodic Table in a very readable way, both for students and teachers ... The number of illustrations is good, and clear. This book is indeed unique and quite thought-provoking ... This book is highly recommended for students, teachers, researchers and not only chemists! Geologists, biochemist and also physicists will find it very interesting to read.' [Read Full Review]Chemistry InternationalThat fossilized chart on every classroom wall — isn't that The Periodic Table? Isn't that what Mendeléev devised about a century ago? No and No. There are many ways of organizing the chemical elements, some of which are thought-provoking, and which reveal philosophical challenges. Where does hydrogen 'belong'? Can an element occupy more than one location on the chart? Which are the Group 3 elements? Is aluminum in the wrong place? Why is silver(I) like thallium(I)? Why is vanadium like molybdenum? Why does gold form an auride ion like a halide ion? Does an atom 'know' if it is a non-metal or metal? Which elements are the 'metalloids'? Which are the triels? So many questions! In this stimulating and innovative book, the Reader will be taken on a voyage from the past to the present to the future of the Periodic Table. This book is unique. This book is readable. This book is thought-provoking. It is a multi-dimensional examination of patterns and trends among the chemical elements. Every reader will discover something about the chemical elements which will provoke thought and a new appreciation as to how the elements relate together.

**The Periodic Table Book** DK,2017-03-30 The Periodic Table Book is the perfect visual guide to the chemical elements that make up our world. This eye-catching encyclopedia takes children on a visual tour of the 118 chemical elements of the periodic table, from argon to zinc. It explores the naturally occurring elements, as well as the man-made ones, and explains their properties and atomic structures. Using more than 1,000 full-colour photographs, The Periodic Table Book shows the many natural forms of each element, as well as a wide range of both everyday and unexpected objects in which it is found, making each element relevant for the child's world.

*The Periodic Law* Albert Edward Garrett,1909

**Animated Science: Periodic Table** John Farndon,2021-10-05 The elements of the periodic table come alive in the first book in a stellar nonfiction comic series illustrated by Shiho Pate! From oxygen to hydrogen, carbon to plutonium, Animated Science: Periodic Table makes chemistry come alive! In this book you'll meet the building blocks of you, the world, and the universe and see how they come together to make everything you see, do, and use every day. With a narrative nonfiction text, kid-friendly information, and Shiho Pate's engaging illustrations, Animated Science: Periodic Table is a perfect introduction and ready reference, appealing and laugh-out-loud funny. Easily accessible for readers just learning the elements, with more interesting facts and details for older kids honing their knowledge. Great for all ages!

**Mendeleev on the Periodic Law** Dmitri Ivanovich Mendeleev,2013-04-25 By the dawn of the nineteenth century, elements had been defined as basic building blocks of nature resistant to decomposition by chemical means. In 1869, the Russian chemist Dmitri Ivanovich Mendeleev organized the discord of the elements into the periodic table, assigning each element to a row, with each row corresponding to an elemental category. The underlying order of matter, hitherto only dimly perceived, was suddenly clearly revealed. This is the first English-language collection of Mendeleev's most important writings on the periodic law. Thirteen papers and essays, divided into three groups, reflect the period corresponding to the initial establishment of the periodic law (three papers: 1869-71), a period of priority disputes and experimental confirmations (five papers: 1871-86), and a final period of general acceptance for the law and increasing international recognition for Mendeleev (five papers: 1887-1905). A single, easily accessible source for Mendeleev's principle papers, this volume offers a history of the development of the periodic law, written by the law's own founder.

**The Periodic Table** Tom Jackson,2020-02-04 Which is the densest element? Which has the largest atoms? And why are some elements radioactive? From the little-known uses of gold in medicine to the development of the hydrogen bomb, this is a fresh new look at the Periodic Table. Combining cutting edge science with fascinating facts and stunning infographics, this book looks at the extraordinary stories of discovery, amazing properties and surprising uses of each elements, whether solid, liquid or gas - naturally occurring, synthesised or theoretical! From hydrogen to oganesson, this is a fact-filled visual guide to each element,each accompanied by technical data (category, atomic number, weight, boiling point) as well as fun facts and stories about their discovery and surprising uses.

**The Periodic Table** Eric R. Scerri,2019 The periodic table of elements, first encountered by many of us at school, provides an arrangement of the chemical elements, ordered by their atomic number, electron configuration, and recurring chemical properties, and divided into periodic trends. In this Very Short Introduction Eric R. Scerri looks at the trends in properties of elements that led to the construction of the table, and shows how the deeper meaning of the table's structure gradually became apparent with the development of atomic theory and, in particular, quantum mechanics, which underlies the behaviour of all of the elements and their compounds. This new edition, publishing in the International Year of the Periodic Table, celebrates the completion of the seventh period of the table, with the ratification and naming of elements 113, 115, 117, and 118 as nihonium, moscovium, tennessine, and oganesson. Eric R. Scerri also incorporates new material on recent advances in our understanding of the origin of the elements, as well as developments concerning group three of the periodic table. ABOUT THE SERIES: The Very Short Introductions series from Oxford University Press contains hundreds of titles in almost every subject area. These pocket-sized books are the perfect way to get ahead in a new subject quickly. Our expert authors combine facts, analysis, perspective, new ideas, and enthusiasm to make interesting and challenging topics highly readable.

**The Periodic Table** Paul Parsons,2014-03-11 As one of the most recognizable images in science, the periodic table is ingrained in our culture. First drawn up in 1869 by Dmitri Mendeleev, its 118 elements make up not only everything on our planet but also everything in the entire universe. The Periodic Table looks at the fascinating story and surprising uses of each of those elements, whether solid, liquid or gas. From the little-known uses of gold in medicine to the development of the hydrogen bomb, each entry is accompanied by technical data (category, atomic number, weight, boiling point) presented in easy-to-read headers, and a colour coding system that helps the reader to navigate through the different groups of elements. A remarkable display of thought-provoking science and beautiful photography, this guide will allow the reader to discover the world afresh.

*Elemental* Tim James,2019-03-26 If you want to understand how our world works, the periodic table holds the answers. When the seventh row of the periodic table of elements was completed in June 2016 with the addition of four final elements—nihonium, moscovium, tennessine, and oganesson—we at last could identify all the ingredients necessary to construct our world.In Elemental,

chemist and science educator Tim James provides an informative, entertaining, and quirkily illustrated guide to the table that shows clearly how this abstract and seemingly jumbled graphic is relevant to our day-to-day lives. James tells the story of the periodic table from its ancient Greek roots, when you could count the number of elements humans were aware of on one hand, to the modern alchemists of the twentieth and twenty-first centuries who have used nuclear chemistry and physics to generate new elements and complete the periodic table. In addition to this, he answers questions such as: What is the chemical symbol for a human? What would happen if all of the elements were mixed together? Which liquid can teleport through walls? Why is the medieval dream of transmuting lead into gold now a reality? Whether you're studying the periodic table for the first time or are simply interested in the fundamental building blocks of the universe—from the core of the sun to the networks in your brain—Elemental is the perfect guide.

*The Periodic Table* Eric R. Scerri, 2019 The periodic table of elements is among the most recognizable image in science. It lies at the core of chemistry and embodies the most fundamental principles of science. In this new edition, Eric Scerri offers readers a complete and updated history and philosophy of the periodic table. Written in a lively style to appeal to experts and interested laypersons alike, *The Periodic Table: Its Story and Its Significance* begins with an overview of the importance of the periodic table and the manner in which the term element has been interpreted by chemists and philosophers across time. The book traces the evolution and development of the periodic table from its early beginnings with the work of the precursors like De Chancourtois, Newlands and Meyer to Mendeleev's 1869 first published table and beyond. Several chapters are devoted to developments in 20th century physics, especially quantum mechanics and the extent to which they explain the periodic table in a more fundamental way. Other chapters examine the formation of the elements, nuclear structure, the discovery of the last seven infra-uranium elements, and the synthesis of trans-uranium elements. Finally, the book considers the many different ways of representing the periodic system and the quest for an optimal arrangement.

**The Periodic Table** Mary Griffin, 2018-12-15 The periodic table is the way scientists have organized the known elements, but up-and-coming scientists can learn much from the periodic table as well. Your readers will learn how to read the periodic table, including what an element's atomic number means and what the rows and columns signify. They'll gather facts about common and interesting elements and even discover how they can predict how elements might chemically react from their position in the table. Fact boxes and images add additional information to the accessible text.

*Periodic Tales* Hugh Aldersey-Williams, 2011-03-29 In the spirit of *A Short History of Nearly Everything* comes *Periodic Tales*. Award-winning science writer Hugh Aldersey-Williams offers readers a captivating look at the elements—and the amazing, little-known stories behind their discoveries. *Periodic Tales* is an energetic and wide-ranging book of innovations and innovators, of superstition and science and the myriad ways the chemical elements are woven into our culture, history, and language. It will delight readers of *Genome*, *Einstein's Dreams*, *Longitude*, and *The Age of Wonder*.

*Early Responses to the Periodic System* Masanori Kaji, Helge Kragh, Gabor Pallo, 2015-01-29 The reception of the periodic system of elements has received little attention among scientists and historians alike. While many historians have studied Mendeleev's discovery of the periodic system, few have analyzed the ways in which the scientific community perceived and employed it. American historian of science Stephen G. Brush concluded that the periodic law had been generally accepted in the United States and Britain, and has suggested the need to extend this study to other countries. In *Early Responses to the Periodic System*, renowned historians of science Masanori Kaji, Helge Kragh, and Gábor Palló present the first major comparative analysis on the reception, response, and appropriation of the periodic system of elements among different nation-states. This book examines the history of its pedagogy and popularization in scientific communities, educational sectors, and popular culture from the 1970s to the 1920s. Fifteen notable historians of science explore the impact of Mendeleev's discovery in eleven countries (and one region) central to chemical research, including Russia, Germany, the Czech lands, and Japan, one of the few nation-states outside the Western world to participate in the nineteenth-century scientific research. The collection, organized by nation-state, explores how local actors regarded the new discovery as law, classification, or theoretical interpretation. In addition to discussing the appropriation of the periodic system, the book examines meta-physical reflections of nature based on the periodic system outside the field of chemistry, and considers how far humans can push the categories of response and reception. *Early Responses to the Periodic System* provides a compelling read for anyone with an interest in the history of chemistry and the Periodic Table of Elements.

*The Periodic Kingdom* P. W. Atkins, 2008-08-05 Come on a journey into the heart of matter--and enjoy the process!--as a brilliant scientist and entertaining tour guide takes you on a fascinating voyage through the Periodic Kingdom, the world of the elements. The periodic table, your map for this trip, is the most important concept in chemistry. It hangs in classrooms and labs throughout the world, providing support for students, suggesting new avenues of research for professionals, succinctly organizing the whole of chemistry. The one hundred or so elements listed in the table make up everything in the universe, from microscopic organisms to distant planets. Just how does the periodic table help us make sense of the world around us? Using vivid imagery, ingenious analogies, and liberal doses of humor P. W. Atkins answers this question. He shows us that the Periodic Kingdom is a systematic place. Detailing the geography, history and governing institutions of this imaginary landscape, he demonstrates how physical similarities can point to deeper affinities, and how the location of an element can be used to predict its properties. Here's an opportunity to discover a rich kingdom of the imagination kingdom of which our own world is a manifestation.

**The Periodic Table I** D. Michael P. Mingos, 2020-02-05 As 2019 has been declared the International Year of the Periodic Table, it is appropriate that Structure and Bonding marks this anniversary with two special volumes. In 1869 Dmitri Ivanovitch Mendeleev first proposed his periodic table of the elements. He is given the major credit for proposing the conceptual framework used by chemists to systematically inter-relate the chemical properties of the elements. However, the concept of periodicity evolved in distinct stages and was the culmination of work by other chemists over several decades. For example, Newland's Law of Octaves marked an important step in the evolution of the periodic system since it represented the first clear statement that the properties of the elements repeated after intervals of 8. Mendeleev's predictions demonstrated in an impressive manner how the periodic table could be used to predict the occurrence and properties of new elements. Not all of his many predictions proved to be valid, but the discovery of scandium, gallium and germanium represented sufficient vindication of its utility and they cemented its enduring influence. Mendeleev's periodic table was based on the atomic weights of the elements and it was another 50 years before Moseley established that it was the atomic number of the elements, that was the fundamental parameter and this led to the prediction of further elements. Some have suggested that the periodic table is one of the most fruitful ideas in modern science and that it is comparable to Darwin's theory of evolution by natural selection, proposed at approximately the same time. There is no doubt that the periodic table occupies a central position in chemistry. In its modern form it is reproduced in most undergraduate inorganic textbooks and is present in almost every chemistry lecture room and classroom. This first volume provides chemists with an account of the historical development of the Periodic Table and an overview of how the Periodic Table has evolved over the last 150 years. It also illustrates how it has guided the research programmes of some distinguished chemists.

**Mendeleev to Oganesson** Eric R. Scerri, Guillermo Restrepo, 2018 An edited volume featuring chapters on multidisciplinary aspects of the Periodic Table, particularly focusing on the history and philosophy of chemistry.

*The Complete Periodic Table* Adrian Dingle,2015 Do you confuse boron with barium or chlorine with fluorine? Fear not! Basher Science has come to the rescue by mixing science and art to create a unique periodic table. From unassuming oxygen to devious manganese, the incredible elements show you the periodic table as you've never seen it before. Basher Science: The Periodic Table gives a face, voice and personality to the chemical elements, making learning chemistry easy and a whole lot more fun.

*The Periodic Table* Becky Ham,2009 A short guide to one of the most celebrated diagrams in the history of science, a tool for understanding the basic building blocks of the universe.

**The Basics of the Periodic Table** Leon Gray,2013-12-15 A sweeping history of both the discovery and classification of elements and the development of the modern periodic table. Included are discussions of the discovery of matter, atoms, atomic structure, molecules, compounds, ions, and isotopes, as well as the first identifications of the 118 (and counting) elements and the various ways they have been classified and organized by prominent scientists up to the present-day periodic table. Instruction in how to read the periodic table is accompanied by examinations of the various groups of elements, their location on the table, and their properties and practical uses. This text strongly supports Common Core Standards for the reading of scientific and technical texts and accounts, and furnishes ample opportunities to summarize, cite evidence, and analyze connections between ideas, individuals, and events.

**Essential Elements** Matt Tweed,2003-04-01 A short, illustrated introduction to the tiny building blocks of our universeincluding atoms, quarks, and the periodic table. Illustrations. 10,000print.

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