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Akitoshi Shiotari

Report ,1912

Report on the Danish Oceanographical Expedition to the Mediterranean and Adjacent Seas ,1912

Wind Waves at Sea, Breakers and Surf Henry Bryant Bigelow, Walles Thomas Edmondson, 1947

Development of a Quality Assurance Program for Asphalt Paving Mixtures in South Carolina J. L. Burati, 2003

Introduction, hydrography, deposits of the sea-bottom Johannes Schmidt, 1912

Surf Statistics for the Coasts of the United States James R. Helle, 1958

Chesapeake Bay Environmental Data Directory Dan Jacobs, 1987

NASA Technical Note ,1969

Revised FORTRAN Program for Calculating Velocities and Streamlines on a Blade to Blade Stream Surface of a Turbomachine Theodore Katsanis, William D. McNally, 1969

Ammonia Anders Nielsen, 2012-12-06 Ammonia is one of the 10 largest commodity chemicals produced. The editor, Anders Nielsen, is research director with one of the largest industrial catalyst producers. He has compiled a complete reference on all aspects of catalytical ammonia production in industry, from thermodynamics and kinetics to reactor and plant design. One chapter deals with safety aspects of ammonia handling and storage.

Hang Ten! Surfing Christine Dugan, 2012-07-01 Find out what surfers need to know to have fun and stay safe in the water in this engaging nonfiction reader! Invite readers to learn about the history of surfing, surfing culture, what equipment is needed, and more using vibrant images and informational text in conjunction with math skills and upper-level features such as chapter format, a glossary of terms, and an index.

Introduction, hydrography, deposits of the sea-bottom ,1912

Simple Processes at the Gas-Solid Interface R.G. Compton, C.H. Bamford, C.F.H. Tippet, 1984-07-01 Simple Processes at the Gas-Solid Interface

Experimental Innovations in Surface Science John T. Yates Jr., 2015-08-17 This book is a new edition of a classic text on experimental methods and instruments in surface science. It offers practical insight useful to chemists, physicists, and materials scientists working in experimental surface science. This enlarged second edition contains almost 300 descriptions of experimental methods. The more than 50 active areas with individual scientific and measurement concepts and activities relevant to each area are presented in this book. The key areas covered are: Vacuum System Technology, Mechanical Fabrication Techniques, Measurement Methods, Thermal Control, Delivery of Adsorbates to Surfaces, UHV Windows, Surface Preparation Methods, High Area Solids, Safety. The book is written for researchers and graduate students.

Material Concepts in Surface Reactivity and Catalysis Henry Wise, 2012-12-02 Material Concepts in Surface Reactivity and Catalysis focuses on the physical and chemical properties of the surface in a reacting system, identifying surface properties that influence interfacial reactions in metal and nonmetal systems. This book discusses the bulk and surface imperfections, surface energy of multicomponent systems, adsorption isotherms and isosteres, and Lennard-Jones potential curves. The adsorbate-induced surface reconstruction, metal-support reactions, defect thermodynamics, and defect metal oxides with crystallographic shear structures are also elaborated. This text likewise covers the metal oxide catalysis, component segregation at grain boundaries, diffusion-controlled metal oxidation, and kinetics of metal deposition. This publication is useful to students in materials science, solid-state chemistry, and catalysis, as well as specialists engaged in research.

Introduction to Surface Chemistry and Catalysis Gabor A. Somorjai, Yimin Li, 2010-06-08 Now updated-the current state of development of modern surface science Since the publication of the first edition of this book, molecular surface chemistry and catalysis science have developed rapidly and expanded into fields where atomic scale and molecular information were previously not available. This revised edition of Introduction to Surface Chemistry and Catalysis reflects this increase of information in virtually every chapter. It emphasizes the modern concepts of surface chemistry and

catalysis uncovered by breakthroughs in molecular-level studies of surfaces over the past three decades while serving as a reference source for data and concepts related to properties of surfaces and interfaces. The book opens with a brief history of the evolution of surface chemistry and reviews the nature of various surfaces and interfaces encountered in everyday life. New research in two crucial areas-nanomaterials and polymer and biopolymer interfaces-is emphasized, while important applications in tribology and catalysis, producing chemicals and fuels with high turnover and selectivity, are addressed. The basic concepts surrounding various properties of surfaces such as structure, thermodynamics, dynamics, electrical properties, and surface chemical bonds are presented. The techniques of atomic and molecular scale studies of surfaces are listed with references to up-to-date review papers. For advanced readers, this book covers recent developments in in-situ surface analysis such as high- pressure scanning tunneling microscopy, ambient pressure X-ray photoelectron spectroscopy, and sum frequency generation vibrational spectroscopy (SFG). Tables listing surface structures and data summarizing the kinetics of catalytic reactions over metal surfaces are also included. New to this edition: A discussion of new physical and chemical properties of nanoparticles Ways to utilize new surface science techniques to study properties of polymers, reaction intermediates, and mobility of atoms and molecules at surfaces Molecular-level studies on the origin of the selectivity for several catalytic reactions A microscopic understanding of mechanical properties of surfaces Updated tables of experimental data A new chapter on soft surfaces, polymers, and biointerfaces Introduction to Surface Chemistry and Catalysis serves as a textbook for undergraduate and graduate students taking advanced courses in physics, chemistry, engineering, and materials science, as well as researchers in surface science, catalysis science, and their applications.

Fundamental Understanding of Mercury Removal from Coal Combustion Erdem

Sasmaz,2011 Coal-fired power plants are a major anthropogenic source of worldwide mercury (Hg) emissions. Since mercury is considered to be one of the most toxic metals found in the environment, Hg emissions from coal-fired power plants is of major environmental concern. Mercury in coal is vaporized into its gaseous elemental form throughout the coal combustion process. Elemental Hg can be oxidized in subsequent reactions with other gaseous components (homogeneous) and solid materials (heterogeneous) in coal-fired flue gases. While oxidized Hg in coal-fired flue gases is readily controlled by its adsorption onto fly ash and/or its dissolution into existing solution-based sulfur dioxide (SO₂) scrubbers, elemental Hg is not controlled. The extent of elemental Hg formed during coal combustion is difficult to predict since it is dependent on the type of coal burned, combustion conditions, and existing control technologies installed. Therefore, it is important to understand heterogeneous Hg reaction mechanisms to predict the speciation of Hg emissions from coal-fired power plants to design and effectively determine the best applicable control technologies. In this work, theoretical and experimental investigations have been performed to investigate the adsorption and in some cases the oxidation, of Hg on solid surfaces, e.g., calcium oxide (CaO), noble metals and activated carbon (AC). The objective of this research is to identify potential materials that can be used as multi-pollutant sorbents in power plants by carrying out both high-level density functional theory (DFT) electronic structure calculations and experiments to understand heterogeneous chemical pathways of Hg. This research uses a fundamental science-based approach to understand the environmental problems caused by coal-fired energy production and provides solutions to the power generation industry for emissions reductions. Understanding the mechanism associated with Hg and SO₂ adsorption on CaO will help to optimize the conditions or material to limit Hg emissions from the flue gas desulfurization process. Plane-wave DFT calculations were used to investigate the binding mechanism of Hg species and SO₂ on the CaO(100) surface. The binding strengths on the high-symmetry CaO adsorption sites have been investigated for elemental Hg, SO₂, mercury chlorides (HgCl and HgCl₂) and mercuric oxide (HgO). It has been discovered that HgCl, HgCl₂, and SO₂ chemisorb on the CaO(100) surface at 0.125 ML coverage. Binding energies of elemental Hg are minimal indicating a physisorption mechanism. Noble metals such as palladium (Pd), gold (Au), silver (Ag), and copper (Cu) have been proposed to capture elemental Hg. Plane-wave DFT calculations

have been carried out to investigate the mercury interactions with Pd binary alloys and overlays in addition to pure Pd, Au, Ag, and Cu surfaces. It has been determined that Pd has the highest mercury binding energy in comparison to other noble metals. In addition, Pd is found to be the primary surface atom responsible for increasing the adsorption of Hg with the surface in both Pd binary alloys and overlays. Deposition of Pd overlays on Au and Ag has been found to enhance the reactivity of the surface by shifting the d-states of surface atoms up in energy. The possible binding mechanisms of elemental Hg onto virgin, brominated and sulfonated AC fiber and brominated powder AC sorbents have been investigated through packed-bed experiments in a stream of air and simulated flue gas conditions, including SO₂, hydrogen chloride (HCl), nitrogen oxide (NO) nitrogen dioxide (NO₂). A combination of spectroscopy and plane-wave DFT calculations was used to characterize the sorption process. X-ray photoelectron spectroscopy (XPS) and x-ray absorption fine structure (XAFS) spectroscopy were used to analyze the surface and bulk chemical compositions of brominated AC sorbents reacted with Hg⁰. Through XPS surface characterization studies it was found that Hg adsorption is primarily associated with halogens on the surface. Elemental Hg is oxidized on AC surfaces and the oxidation state of adsorbed Hg is found to be Hg²⁺. Though plane-wave DFT and density of states (DOS) calculations indicate that Hg is more stable when it is bound to the edge carbon atom interacting with a single bromine bound atop of Hg, a model that includes an interaction between the Hg and an additional Br atom matches best with experimental data obtained from extended x-ray absorption fine structure (EXAFS) spectroscopy. The flue gas species such as HCl and bromine (Br₂) enhance the Hg adsorption, while SO₂ is found to decrease the Hg adsorption significantly by poisoning the active sites on the AC surface. The AC sorbents represent the most market-ready technology for Hg capture and therefore have been investigated by both theory and experiment in this work. Future work will include similar characterization and bench-scale experiments to test the metal-based materials for the sorbent and oxidation performance.

Report on the Danish Oceanographical Expeditions 1908-1910 to the Mediterranean and Adjacent Seas Johannes Schmidt, 1912

Bulletin, 1928

Reactivity of Nitric Oxide on Copper Surfaces Akitoshi Shiotari, 2017-06-07 This book provides microscopic insights into chemical properties of NO on metal surfaces. NO/metal systems have been studied intensively to understand heterogeneous catalysis to detox exhaust NO_x gas. The identification and componential analysis of various and mixed chemical species of NO adsorbed onto the surfaces have been significant challenges faced by conventional experimental techniques, such as vibrational spectroscopies. The author investigated individual NO molecules on Cu surfaces using low-temperature scanning tunneling microscopy (STM). STM not only provides information on the geometric, electronic, and vibrational properties at the single-molecule level; it is also able to manipulate molecules on surfaces to induce chemical reaction. Exploiting those techniques, the author chemically identified individual NO-related species on the surfaces and discovered new reaction processes for NO reduction, which provides microscopic insights into the catalytic mechanisms. The author also visualized wave functions of electrons in a valence orbital of NO and demonstrated that the wave functions are modified by the formation of covalent bonding or hydrogen bonding. This is, namely, the visualization of quantum mechanics in real space, which is certainly worth reading. Furthermore, the book demonstrates that direct observation of valence orbitals helps to elucidate the reactivity of molecules adsorbed onto surfaces. This innovative approach to studying molecular properties will contribute to further development of STM and its related methods.

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