

PURE GAS

RALPH HIGBIE

1. **THE PRESSURE-INDUCED FUNDAMENTAL BAND OF OXYGEN IN PURE GAS AND IN OXYGEN-FOREIGN GAS MIXTURES AT 298°K AND AT 201°K** ROBERT B. BISHOP, ROBERT BRADLEY, 1966

2. **GAS SEPARATION BY ADSORPTION PROCESSES** RALPH T. YANG, 2013-10-22 GAS SEPARATION BY ADSORPTION PROCESSES PROVIDES A THOROUGH DISCUSSION OF THE ADVANCEMENT IN GAS ADSORPTION PROCESS. THE BOOK IS COMPRISED OF EIGHT CHAPTERS THAT EMPHASIZE THE FUNDAMENTALS CONCEPT AND PRINCIPLES. THE TEXT FIRST COVERS THE ADSORBENTS AND ADSORPTION ISOTHERMS, AND THEN PROCEEDS TO DETAILING THE EQUILIBRIUM ADSORPTION OF GAS MIXTURES. NEXT, THE BOOK COVERS RATE PROCESSES IN ADSORBERS AND ADSORBER DYNAMICS. THE NEXT CHAPTER DISCUSSES CYCLIC GAS SEPARATION PROCESSES, AND THE REMAINING TWO CHAPTERS COVER PRESSURE-SWING ADSORPTION. THE BOOK WILL BE OF GREAT USE TO STUDENTS, RESEARCHERS, AND PRACTITIONERS OF DISCIPLINES THAT INVOLVE GAS SEPARATION PROCESSES, SUCH AS CHEMICAL ENGINEERING.

3. **PRODUCTION OF WELL KNOWN PRESSURES OF PURE AND MIXED GASES IN THE HIGH-VACUUM RANGE** G. SCHNEPPE, 1973 A GAS INLET SYSTEM IS DESCRIBED WHICH USES THE PRESSURE-BASED DYNAMIC EXPANSION METHOD TO PRODUCE WELL-KNOWN PRESSURES OVER THE RANGE FROM 1/(10 TO THE 4TH) TO LESS THAN 1/(10 TO THE 8TH) TORR. THE AVERAGE ABSOLUTE ACCURACY IS ABOUT 2%, AND THE REPRODUCIBILITY BETTER THAN 1%. ADJUSTMENTS OF THE TEST PRESSURE ARE PERFORMED WITHIN A FEW SECONDS. THE GAS IS INTRODUCED INTO THE TEST VOLUME THROUGH A POROUS GLASS FILTER. ITS SMALL CONDUCTANCE CAUSES A STRONG PRESSURE REDUCTION, SO THAT THE REFERENCE PRESSURE CAN BE MEASURED BY A COMPENSATED CAPACITY MANOMETER, WHICH IN TURN IS CALIBRATED AGAINST AN HG-U-MANOMETER. THIS TECHNIQUE IS SUITABLE ALSO FOR MEASUREMENTS OF ACTIVE AND CORROSIVE GASES. IN ADDITION, THE APPARATUS DESCRIBED CONSISTS OF TWO INDEPENDENT GAS INLET SYSTEMS WHICH ALLOW THEM TO PRODUCE GAS MIXTURES AS QUICKLY AND ACCURATELY AS A PURE GAS PRESSURE. IT IS USEFUL FOR MASS SPECTROMETER CALIBRATION.

4. **GAS ADSORPTION EQUILIBRIA** JÜRGEN U. KELLER, REINER STAUDT, 2006-06-23 THIS BOOK IS INTENDED TO PRESENT FOR THE FIRST TIME EXPERIMENTAL METHODS TO MEASURE EQUILIBRIA STATES OF PURE AND MIXED GASES BEING ADSORBED ON THE SURFACE OF SOLID MATERIALS. IT HAS BEEN WRITTEN FOR ENGINEERS AND SCIENTISTS FROM INDUSTRY AND ACADEMIA WHO ARE INTERESTED IN ADSORPTION BASED GAS SEPARATION PROCESSES AND/OR IN USING GAS ADSORPTION FOR CHARACTERIZATION OF THE POROSITY OF SOLID MATERIALS. THIS BOOK IS THE RESULT OF A FRUITFUL COLLABORATION OF A THEORETICIAN (JUK) AND AN EXPERIMENTALIST (RS) OVER MORE THAN TWELVE YEARS IN THE FIELD OF GAS ADSORPTION SYSTEMS AT THE INSTITUTE OF FLUID- AND THERMODYNAMICS (IFT) AT THE UNIVERSITY OF SIEGEN, SIEGEN, GERMANY. THIS COLLABORATION RESULTED IN THE DEVELOPMENT OF SEVERAL NEW METHODS TO MEASURE NOT ONLY PURE GAS ADSORPTION, BUT GAS MIXTURE OR COADSORPTION EQUILIBRIA ON INERT POROUS SOLIDS. ALSO SEVERAL NEW THEORETICAL RESULTS COULD BE ACHIEVED LEADING TO NEW TYPES OF SO-CALLED ADSORPTION ISOTHERMS BASED ON THE CONCEPTS OF MOLECULAR ASSOCIATION AND – PHENOMENOLOGICALLY SPEAKING – ON THAT OF THERMODYNAMIC PHASES OF FRACTAL DIMENSION. NATURALLY, RESULTS OF INTERNATIONAL COLLABORATION OF THE AUTHORS OVER THE YEARS (1980-2000) ALSO ARE INCLUDED.

5. **THE PRESSURE-INDUCED FUNDAMENTAL BAND OF OXYGEN IN PURE GAS AND IN OXYGEN-FOREIGN GAS MIXTURES AT 298°K AND AT 201°K**. ROBERT B. BISHOP, 1966

6. **CALCULATION OF TRANSPORT PROPERTIES OF IONIZING ATOMIC HYDROGEN** NORMAN T. GRIER, 1966

7. **A PURPOSE BUILT PURE GAS TRANSFER AND PULSE MEASUREMENT SYSTEM TO COLLECT DATA ON THE PERFORMANCE OF GAS FILLINGS FOR HIGH TEMPERATURE PULSE FISSION CHAMBERS** S. BYROM, 1985

8. **THE RATE OF ABSORPTION OF A PURE GAS INTO STILL LIQUID DURING SHORT PERIODS OF EXPOSURE** RALPH HIGBIE, 1935

9. **SOME EXPERIMENTS ON THE INITIATION OF COAL-DUST EXPLOSIONS BY GAS EXPLOSIONS** GEORGE SAMUEL RICE, HAROLD PUTNAM GREENWALD, HENRY C. HOWARTH, 1930

10. **GAS TRANSPORT IN GLASSY POLYMERS** MARIA GRAZIA DE ANGELIS, GIULIO C. SARTI, 2021-04-22 THIS SPECIAL ISSUE OF MEMBRANES FOCUSES ON SEVERAL NEW ASPECTS OF FLUID TRANSPORT IN GLASSY POLYMERS, WITH APPLICATION IN RELEVANT MEMBRANE SEPARATIONS SUCH AS GAS PURIFICATION, VOC REMOVAL AND CO2 CAPTURE. IN PARTICULAR, THE FOCUS LIES ON NOVEL EXPERIMENTAL TECHNIQUES, AND DETAILED CHARACTERIZATION OF SPECIFIC PHENOMENA LIKE POLAR AND MULTICOMPONENT INTERACTIONS DURING TRANSPORT. THE PROPERTIES OF NOVEL MATERIALS, SUCH AS MIXED MATRIX MEMBRANES BASED ON GLASSY POLYMERS AND DIFFERENT SELECTIVE FILLERS, ARE ALSO PRESENTED. A CRITICAL REVIEW OF EXISTING MODELING APPROACHES TO DESCRIBE THE SORPTION AND TRANSPORT IN GLASSY POLYMERS SUITABLE FOR MEMBRANE SEPARATIONS IS PROVIDED, INCLUDING BOTH MACROSCOPIC AND ATOMISTIC MODELS, AND RELYING BOTH ON THE STANDARD SOLUTION-DIFFUSION PROCESS AND ON THE FACILITATED TRANSPORT MECHANISM.

11. **DESIGN STUDY REPORT FOR TARGET** GENERAL DYNAMICS CORPORATION. GENERAL ATOMIC DIVISION. JOHN JAY HOPKINS LABORATORY FOR PURE AND APPLIED SCIENCE, 1964

12. **THERMAL CONDUCTIVITY OF GAS MIXTURES** ALEXANDER LEWIS LINDSAY, LEROY A. BROMLEY, 1949

13. **SEPARATION OF A GAS MIXTURE FLOWING THROUGH A LONG TUBE AT LOW PRESSURE** RICHARD DAVID PRESENT, 1949 THE SEPARATION OF A BINARY GAS MIXTURE BY DIFFUSION THROUGH A CAPILLARY OF RADIUS R DEPENDS ON THE FACT THAT THE MOLECULES HAVE DIFFERENT MASSES $m_{sub\ i}$ AND MEAN SPEEDS $v_{sub\ i}$. WHEN THE INLET PRESSURE IS SO LOW THAT THE MEAN FREE PATH λ IS MUCH GREATER THAN R, THE FLOW IS DIFFUSIVE AND THE SEPARATION FACTOR (AT ZERO OUTLET PRESSURE) HAS ITS MAXIMUM VALUE $(m_{sub\ 1}/m_{sub\ 2})^{1/2}$. AT HIGH PRESSURES (λ LESS THAN OR EQUAL R) NO SEPARATION OCCURS. THIS PAPER TREATS THE INTERMEDIATE CASE (λ APPROX. R) WHERE THE TRANSFER OF FORWARD MOMENTUM FROM LIGHT TO HEAVY MOLECULES IN UNLIKE COLLISIONS EQUALIZES THE TRANSPORT VELOCITIES AND DECREASES THE SEPARATION FACTOR. AS THE INLET PRESSURE RISES, THIS EFFECT MAKES THE FLOW NONSEPARATIVE BEFORE IT BECOMES VISCOUS. FLOW EQUATIONS ARE DERIVED BY EQUATING THE MOMENTUM ACQUIRED BY THE LIGHT COMPONENT FROM THE PRESSURE GRADIENT TO THE MOMENTUM LOST TO THE WALL PLUS THAT TRANSFERRED TO THE OTHER COMPONENT. THE VISCOUS EFFECTS ARE TREATED AS A SMALL ADDITIVE PERTURBATION ON THE FLOW. THE INTEGRATED FLOW EQUATIONS EXPRESS THE SEPARATION FACTOR AS A FUNCTION OF THE INLET AND OUTLET PRESSURES.

14. **OIL & GAS JOURNAL**, 1926

15. **THE GREAT CENTRAL GAS COMPANY, ITS ORIGIN AND HISTORY. A PAPER READ BEFORE THE SOCIETY OF ARTS, ETC** ALEXANDER ANGUS CROLL, 1875

16. **HANDBOOK OF LASERS** MARVIN J. WEBER, 2019-04-30 LASERS CONTINUE TO BE AN AMAZINGLY ROBUST FIELD OF ACTIVITY. ANYONE SEEKING A PHOTON SOURCE IS NOW CONFRONTED WITH AN ENORMOUS NUMBER OF POSSIBLE LASERS AND LASER WAVELENGTHS TO CHOOSE FROM, BUT NO SINGLE, COMPREHENSIVE SOURCE TO HELP THEM MAKE THAT CHOICE. THE HANDBOOK OF LASERS PROVIDES AN AUTHORITATIVE COMPILATION OF LASERS, THEIR PROPERTIES, AND ORIGINAL REFERENCES IN A READILY ACCESSIBLE FORM. ORGANIZED BY LASING MEDIA—SOLIDS, LIQUIDS, AND GASES—EACH SECTION IS SUBDIVIDED INTO DISTINCT LASER TYPES. EACH TYPE CARRIES A BRIEF DESCRIPTION, FOLLOWED BY TABLES LISTING THE LASING ELEMENT OR MEDIUM, HOST, LASING TRANSITION AND WAVELENGTH, OPERATING PROPERTIES, PRIMARY LITERATURE CITATIONS, AND, FOR BROADBAND LASERS, REPORTED TUNING RANGES. THE IMPORTANCE AND VALUE OF THE HANDBOOK OF LASERS CANNOT BE OVERSTATED. SERVING AS BOTH AN ARCHIVE AND AS AN INDICATOR OF EMERGING TRENDS, IT REFLECTS THE STATE OF KNOWLEDGE AND DEVELOPMENT IN THE FIELD, PROVIDES A RAPID MEANS OF OBTAINING REFERENCE DATA, AND OFFERS A PATHWAY TO THE LITERATURE. IT CONTAINS DATA USEFUL FOR COMPARISON WITH PREDICTIONS AND FOR DEVELOPING MODELS OF PROCESSES, AND MAY REVEAL FUNDAMENTAL INCONSISTENCIES OR CONFLICTS IN THE DATA.

17. **ON THE NATURE AND CHEMICAL PROPERTIES OF COAL GAS, CONSIDERED WITH RESPECT TO ITS ECONOMICAL AND SANITARY APPLICATION FOR THE PURPOSES OF GENERATING LIGHT AND HEAT** LEWIS THOMPSON, 1851

18. **CURRENT TRENDS AND FUTURE DEVELOPMENTS ON (BIO-) MEMBRANES** ANGELO BASILE, EVANGELOS P. FAVVAS, 2018-07-18 CURRENT TRENDS AND FUTURE DEVELOPMENTS ON (BIO-) MEMBRANES: CARBON DIOXIDE SEPARATION/CAPTURE BY USING MEMBRANES EXPLORES THE UNIQUE PROPERTY OF MEMBRANES TO SEPARATE GASES WITH DIFFERENT PHYSICAL AND CHEMICAL PROPERTIES. THE BOOK COVERS BOTH POLYMERIC AND INORGANIC MATERIALS FOR CO2 SEPARATION AND EXPLAINS THEIR MECHANISM OF ACTION, ALLOWING FOR THE DEVELOPMENT AND MOST APPROPRIATE AND EFFICIENT PROCESSES. IT ALSO LISTS THE ADVANTAGES OF USING MEMBRANES INSTEAD OF OTHER SEPARATION TECHNIQUES, I.E., THEIR LOW OPERATING COSTS AND LOW ENERGY CONSUMPTION. THIS BOOK OFFERS A UNIQUE OPPORTUNITY FOR SCIENTISTS WORKING IN THE FIELD OF MEMBRANE TECHNOLOGY FOR CO2 SEPARATION AND CAPTURE. OUTLINES NUMEROUS MEMBRANE-BASED TECHNOLOGIES FOR CO2 SEPARATION AND CAPTURE LISTS NEW, ADVANCED SEPARATION TECHNIQUES AND PRODUCTION PROCESSES INCLUDES VARIOUS APPLICATIONS, MODELLING, AND THE ECONOMIC CONSIDERATIONS OF EACH PROCESS COVERS ADVANCED TECHNIQUES FOR THE SEPARATION OF CO2 IN NATURAL GAS

19. **GAS AGE**, 1901 INCLUDES SUMMARIES OF PROCEEDINGS AND ADDRESSES OF ANNUAL MEETINGS OF VARIOUS GAS ASSOCIATIONS. L.C. SET INCLUDES AN INDEX TO THESE PROCEEDINGS, 1884-1902, ISSUED AS A SUPPLEMENT TO PROGRESSIVE AGE, FEB. 15, 1910.

20. **THERMAL CONDUCTIVITIES OF ORDINARY AND ISOTOPICALLY SUBSTITUTED POLAR GASES AND THEIR EQUIMOLAR MIXTURES** CHARLES EDWARD BAKER, RICHARD S. BROKAW, 1966

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