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Annelies Wilder-Smith

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that occur for each of the following types of radioactive emissions by determining the algebraic expressions using a or z that would be placed in the boxes for the mass number and the atomic number that would complete the nuclear equation
 β^- decay: ${}^A_Z\text{X} \rightarrow {}^A_{Z+1}\text{Y} + {}^0_{-1}\text{e} + \bar{\nu}$
 β^+ decay: ${}^A_Z\text{X} \rightarrow {}^A_{Z-1}\text{Y} + {}^0_{+1}\text{e} + \nu$
 positron: ${}^A_Z\text{X} \rightarrow {}^A_{Z-1}\text{Y} + {}^0_{+1}\text{e} + \nu$

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web chapter 22 nuclear chemistry nuclear symbols mass number p no 235 92 u atomic number number of p element symbol types of radioactive decay alpha production a helium nucleus ${}^4_2\text{He}$ ${}^{238}_{92}\text{U} \rightarrow {}^{234}_{90}\text{Th} + {}^4_2\text{He}$ beta production ${}^{234}_{91}\text{Pa} \rightarrow {}^{234}_{92}\text{U} + {}^0_{-1}\text{e}$ alpha radiation limited to very large nuclei

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