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Mortimer Levering

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ponyboy curtis belongs

to a lower class group

of oklahoma youths who

call themselves greasers

because of their greasy

long hair walking home

from a movie ponyboy is

attacked by a group of

socs the greasers rivals

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impressive there is not much else to say on that

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of fifteen inspired by
her frustration with the
social divisions in her
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of realistic fiction for
high school readers
explore an in depth
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curtis the plot summary
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s e hinton full name
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attended will rogers
high school and
graduated in 1966 while
she was a student there
she wrote her first book
the outsiders which was
published in 1967
following the suggestion
of her editor she wrote

it under the name s e
 hinton in order to
 deflect attention from
 her gender
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 outsiders is about two
 weeks in the life of a
 14 year old boy the
 novel tells the story of
 ponyboy curtis and his
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following reactions and
identify each type of
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 $\text{CaBr}_2 + \text{NaOH} + \text{NH}_3 + \text{H}_2 + \text{SO}_4$
 $\text{NH}_4^+ + \text{SO}_4^{2-} + \text{C}_5\text{H}_9\text{O} + \text{O}_2 + \text{CO}_2$
 $\text{H}_2\text{O} + \text{Pb} + \text{H}_3\text{PO}_4 + \text{H}_2 + \text{Pb}_3\text{PO}_4$
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 1 $\text{Na}_3\text{PO}_4 + 3\text{KOH} \rightarrow 3\text{NaOH} + \text{K}_3\text{PO}_4$ double displacement
 2 $\text{MgCl}_2 + \text{Li}_2\text{CO}_3 \rightarrow \text{MgCO}_3 + 2\text{LiCl}$ double displacement
 3 $\text{C}_6\text{H}_{12} + 9\text{O}_2 \rightarrow 6\text{CO}_2 + 6\text{H}_2\text{O}$ combustion
 4 $\text{Pb} + \text{FeSO}_4 \rightarrow \text{PbSO}_4 + \text{Fe}$ single displacement
 5 $\text{CaCO}_3 \rightarrow \text{CaO} + \text{CO}_2$ decomposition
 6 $\text{P}_4 + 3\text{O}_2 \rightarrow 2\text{P}_2\text{O}_3$
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equations can also be used to represent physical processes write a chemical reaction for the freezing of water including the proper phase labels explain why
 $4\text{Na} + \text{S} + 2\text{Cl}_2 \rightarrow \text{g}$

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 web 1 3nabr 1h3po41na3po4 3hbr type of reaction double displacement 2 3ca oh 2 1al2 so4 33caso4 2al oh 3 type of reaction double displacement 3 3mg 1fe2o32fe 3mgo type of reaction single displacement 4 1c2h4 3o22co2 2h2o type of reaction combustion 5

2pbso42pbso3 1o2 type of reaction

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 web a chemical reaction is a process in which one or more substances are converted to one or more different substances the starting substances are called the reactants and the new substances that form are called the products
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chemical change physical
does not change the
identity of the matter
can be changed back
easily chemical new
substance is created
with new physical and
chemical

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