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Alkene Polymerization Reactions with Transition Metal Catalysts Yury Kissin, 2008-03-10 During the past 30 years, the field of alkene polymerization over transition metal catalysts underwent several major changes: 1. The list of commercial heterogeneous Ziegler-Natta catalysts for the synthesis of polyethylene and stereoregular polyolefins was completely renewed affording an unprecedented degree of control over the polymer structure. 2. Research devoted to metallocene and other soluble transition-metal catalysis has vastly expanded and has shifted toward complexes of transition metals with multidentate ligands. 3. Recent developments in gel permeation chromatography, temperature-rising fractionation, and crystallization fractionation provided the first reliable information about differences between various active centers in transition-metal catalysts. 4. A rapid development of high-resolution ¹³C NMR spectroscopy resulted in greatly expanded understanding of the chemical and steric features of polyolefins and alkene copolymers. These developments require a new review of all aspects of alkene polymerization reactions with transition-metal catalysts. The first chapter in the book is an introductory text for researchers who are entering the field. It describes the basic principles of polymerization reactions with transition-metal catalysts, the types of catalysts, and commercially manufactured polyolefins. The next chapter addresses the principal issue of alkene polymerization catalysis: the existence of catalyst systems with single and multiple types of active centers. The subsequent chapters are devoted to chemistry and stereochemistry of elemental reaction steps, structures of catalyst precursors and reactions leading to the formation of active centers, kinetics of polymerization reactions, and their mechanisms. The book describes the latest commercial polymerization catalysts for the synthesis of polyethylenes and polypropylene. The book provides a detailed description of the multi-center nature of commercial Ziegler-Natta catalysts. The book devotes specialized chapters to the most important aspects of transition metal polymerization catalysts: the reactions leading to the formation of active centers, the chemistry and stereochemistry of elemental polymerization steps, reaction kinetics, and the polymerization mechanism. The book contains an introductory chapter for researchers who are entering the field of polymerization catalysis. It describes the basic principles of polymerization reactions with transition-metal catalysts and the types of commercially manufactured polyolefins and copolymers. The book contains over 2000 references, the most recent up to end of 2006.

CHEMICAL PROCESS CALCULATIONS PRASAD, RAM, 2022-04-13 The present textbook is written for undergraduate students of chemical engineering as per the syllabus framed by AICTE curriculum. It explains the basic chemical process principles in a lucid manner. SI units, chemical stoichiometry and measures of composition, behaviour of gases, vapour pressure of pure substances, and humidity and saturation are covered in detail. In addition, mass and energy balances of chemical processes have also been described. Chemical processes without chemical reactions include fluid flow, mixing, evaporation distillation, absorption and stripping, liquid-liquid extraction, leaching and washing, adsorption, drying, crystallization and membrane separation process. **SALIENT FEATURES**

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Chemistry³ Andrew Burrows, John Holman, Andrew Parsons, Gwen Pilling, Gareth Price, 2013-03-21 Providing equal coverage of organic, inorganic and physical chemistry - coverage that is uniformly authoritative - this text builds on what students may already know and tackles their misunderstandings and misconceptions. The authors achieve unrivalled accessibility through carefully-worded explanations, the introduction of concepts in a logical and progressive manner, and the use of annotated diagrams and step-by-step worked examples. Students are encouraged to engage with the text and appreciate the central role that chemistry plays in our lives through the unique use of real-world examples and visuals. Frequent cross-references highlight the connections between each strand of chemistry and explain the relationship between the topics, so students can develop an understanding of the subject as a whole.

The Rotarian, 1990-11 Established in 1911, The Rotarian is the official magazine of Rotary International and is circulated worldwide. Each issue contains feature articles, columns, and departments about, or of interest to, Rotarians. Seventeen Nobel Prize winners and 19 Pulitzer Prize winners – from Mahatma Gandhi to Kurt Vonnegut Jr. – have written for the magazine.

Concise Chemical Thermodynamics A.P.H. Peters, 2010-07-02 The first two editions of Concise Chemical Thermodynamics proved to be a very popular introduction to a subject many undergraduate students perceive to be difficult due to the underlying mathematics. With its concise explanations and clear examples, the text has for the past 40 years clarified for countless students one of the most complicated branches

Concise Physical Chemistry Donald W. Rogers, 2011-03-31 This book is a physical chemistry textbook that presents the essentials of physical chemistry as a logical sequence from its most

modest beginning to contemporary research topics. Many books currently on the market focus on the problem sets with a cursory treatment of the conceptual background and theoretical material, whereas this book is concerned only with the conceptual development of the subject. Comprised of 19 chapters, the book will address ideal gas laws, real gases, the thermodynamics of simple systems, thermochemistry, entropy and the second law, the Gibbs free energy, equilibrium, statistical approaches to thermodynamics, the phase rule, chemical kinetics, liquids and solids, solution chemistry, conductivity, electrochemical cells, atomic theory, wave mechanics of simple systems, molecular orbital theory, experimental determination of molecular structure, and photochemistry and the theory of chemical kinetics.

Chemical Calculations Paul C. Yates, 2023-04-21 Uniquely organized by chemical rather than mathematical topics, this book relates each mathematical technique to the chemical concepts where it applies. The new edition features additional, revised, and updated material in every chapter and maintains the clarity of the previous edition with the appropriate organization of topics and improved cross-referencing where mathematical techniques occur more than once. The text contains additional worked examples and end-of-chapter exercises with detailed solutions giving students the opportunity to apply previously introduced techniques to chemically related problems. It is an ideal course companion for chemistry courses throughout the length of a degree. Features ■ This book covers the difficult area of mathematics in an easy-to-read format for students and professionals in chemistry and related subjects. ■ Structured according to chemical rather than mathematical topics. ■ Each topic has at least 12 end of chapter applied chemistry problems to provide practice in applying the techniques to real chemistry. ■ Indexing of material by both chemical and mathematical topics. ■ Extends its utility as a concise and practical reference for professionals in a wide array of scientific disciplines involving chemistry.

Advances in Citrus Nutrition Anoop Kumar Srivastava, 2012-05-30 This is a solitary attempt to streamline all the possible information related to citrus nutrition, with emphasis on diagnosis and management of nutrient constraints, employing a variety of state-of-art techniques evolved globally over the years. While doing so care has been taken to include peripheral disciplines so that the discussion becomes more lively and authoritative. An entire array of exclusive subjects has been nicely portrayed with the help of latest data and photographs.

Physical and Chemical Equilibrium for Chemical Engineers Noel de Nevers, 2012-04-25 This book concentrates on the topic of physical and chemical equilibrium. Using the simplest mathematics along with numerous numerical examples it accurately and rigorously covers physical and chemical equilibrium in depth and detail. It continues to cover the topics found in the first edition however numerous updates have been made including: Changes in naming and notation (the first edition used the traditional names for the Gibbs Free Energy and for Partial Molal Properties, this edition uses the more popular Gibbs Energy and Partial Molar Properties,) changes in symbols (the first edition used the Lewis-Randall fugacity rule and the popular symbol for the same quantity, this edition only uses the popular notation,) and new problems have been added to the text. Finally the second edition includes an appendix about the Bridgman table and its use.

Human Exposure to Chlorinated Paraffins in Scandinavia Yuan, Bo, Hui Tay, Joo, de Wit, Cynthia A., Papadopoulou, Eleni, Padilla-Sánchez, Juan Antonio, Småstuen Haug, Line, 2023-04-17 Chlorinated paraffins (CPs) have received global attention by the authorities as well as the industries using or producing CPs. In late 2017, a group of short chain chlorinated paraffins (SCCPs) and replacement CPs were found in both Norwegian and Swedish mothers' milk and, later, in mothers' milk from over 50 countries across the world. This project focused on a comprehensive understanding of body burden and multiple human exposure pathways to both legacy CPs and current-use CPs in a Norwegian cohort. The cohort provided a rare opportunity to explore human exposure to CPs compared to the other consumer chemical contaminants. Combining both internal and external information, it was studied how much and to what extent people are exposed to the chemicals via different pathways, and what the differences are between the legacy CPs and the current CPs in human exposure and accumulation.

Chemical Calculations Paul Yates, 1997-06-05 Chemical Calculations provides an introduction to the mathematics required for physical chemistry courses. This book is unique in that it provides a gentle introduction with a chemistry centered - rather than math centered - approach. Written by a chemist for undergraduate students, it imparts an understanding of the subject from a chemist's viewpoint using examples from real chemistry. It includes illustrations that show exactly how to use calculators to work problems and examples of important chemical problems with fully worked solutions. This book is an ideal companion throughout a chemistry course that can be consulted when required, and used to keep one step ahead of the lecture.

Principles of Physical Chemistry E. Kirk Roberts, 1984

An Introduction To Chemical Thermodynamics R P Rastogi, 2009-11-01 □ Calculations approach: Strong mathematical rigor has been applied, and a complementary physical treatment given, to make students strong in the applied aspects of thermodynamics □ Problem solving presentation: 195 solved examples and 269 unsolved problems have been given. Hints to difficult problems have been given too. □ Concept checking Review Questions have been given at the end of every chapter □ Coverage on thermodynamic discussion of eutectics, solid solutions and phase separation

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Physical Chemistry for the Biosciences Raymond Chang, 2005-02-11 Physical Chemistry for the Biosciences has been optimized for a one-semester introductory course in physical chemistry for students of biosciences.

Twelfth Report, National Spray Model and Application Technology Working Group Meeting Patricia J. Skyler, 2002

Canadian Journal of Chemistry, 1986

THERMODYNAMICS R. C. SRIVASTAVA, SUBIT K. SAHA, ABHAY K. JAIN, 2007-08-07 Though thermodynamics is a tool used in all sciences and technologies, this book is especially designed to acquaint science students with the whole breadth of the subject covering both equilibrium and non-equilibrium regions. Equilibrium thermodynamics covered in the first-seven chapters caters to the needs of students up to the B.Sc./B.Sc. (Hons.) level. The next three chapters devoted to non-equilibrium thermodynamics and network thermodynamics fulfill the needs of the syllabi on these topics introduced in most universities at the postgraduate level. Chapters on 'The Question of Ideality' and 'The Non-linear Region' were the new additions to the second edition. In the third edition a new chapter on "Causality Principle in Non-equilibrium Thermodynamics" has been added. The readers may find the new chapter intellectually stimulating. The book is an accessible, straightforward discussion of basic topics, beginning with the laws of thermodynamics and focusing on derivations of basic relations. The text is suitably illustrated throughout with examples of various applications of interest to science students. It explains concepts systematically, teaches problem-solving meaningfully, and includes concept-elucidating questions that are intended to reinforce the student's understanding of the material.

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