

# **Chemical Elements Reference**

**David E. Newton**

**The History and Use of Our Earth's Chemical Elements** Robert E. Krebs,1998-01-26 Descriptions of all elements, usually only found by consulting many different sources. Includes history and common uses.

**The History and Use of Our Earth's Chemical Elements** Robert E. Krebs,2006-07-30 Understanding the Periodic Table of Chemical Elements is critical for success in the chemistry classroom and laboratory. In today's classroom, students not only need to understand the properties of the chemical elements, but how these elements play such an integral role in industry, the earth and the environment, and in modern life. No resource provides a better introduction than Robert Krebs's The History and Use of Our Earth's Chemical Elements. In this thoroughly revised edition, with extensive new examples on the importance of the chemical elements, the elements are examined within their groups, enabling students to make connections between elements of similar structure. In addition, the discovery and history of each element - from those known from ancient times to those created in the modern laboratory - is explained clearly and concisely. Understanding the Periodic Table of Chemical Elements is critical for success in the chemistry classroom and laboratory. In today's classroom, students not only need to understand the properties of the chemical elements, but how these elements play such an integral role in industry, the earth and the environment, and in modern life. No resource provides a better introduction than Robert Krebs's The History and Use of Our Earth's Chemical Elements. In this thoroughly revised edition, with extensive new and updated examples on the use of the chemical elements, the elements are examined within their groups, enabling students to make connections between elements of similar structure. In addition, the discovery and history of each element - from those known from ancient times to those created in the modern laboratory - is explained clearly and concisely. In addition to the handy Guide to the Chemical Elements that comprises the bulk of the work, The History and Use of Our Earth's Chemical Elements includes other useful features: ; Introductory material on the basics of chemistry and the Periodic Table ; Appendices on the discoverers of the chemical elements ; A glossary of words commonly used in chemistry and chemical engineering ; A complete bibliography of useful resources, including websites All of this information makes The History and Use of Our Earth's Chemical Elements the ideal one-volume resource for understanding the importance of the chemical elements.

*Chemical Elements Pocket Guide* Coventry House Publishing,2020-02-08 The Chemical Elements Pocket Guide serves as a portable reference for quick study and efficient review of the 118 elements on the periodic table. This on-the-go resource details the physical and atomic properties of each element, as well as their history and characteristics in bullet point format. The book's small trim size (4.25 x 6.8 inches) is intended to fit inside a lab coat pocket, and the bound design means you no longer need to carry loose, bulky flashcards that can be misplaced or destroyed. Includes the updated names nihonium, moscovium, tennessine and oganesson for elements 113, 115, 117, and 118, respectively. Information provided includes: • Atomic number • Atomic symbol • Element category • Standard state • Atomic mass • Electron configuration • Oxidation states • Electronegativity • Atomic radius • Ionization energy • Electron affinity • Melting point • Boiling point • Density • Year discovered • Discovered by • Appearance • Natural occurrence • Interesting fact

**Chemical Elements:from Carbon to Krypton** David E. Newton,1999

Chemistry of the Elements N. N. Greenwood,A. Earnshaw,2012-12-02 When this innovative textbook first appeared in 1984 it rapidly became a great success throughout the world and has already been translated into several European and Asian languages.

Now the authors have completely revised and updated the text, including more than 2000 new literature references to work published since the first edition. No page has been left unaltered but the novel features which proved so attractive have been retained. The book presents a balanced, coherent and comprehensive account of the chemistry of the elements for both undergraduate and postgraduate students. This crucial central area of chemistry is full of ingenious experiments, intriguing compounds and exciting new discoveries. The authors specifically avoid the term 'inorganic chemistry' since this evokes an outmoded view of chemistry which is no longer appropriate in the final decade of the 20th century. Accordingly, the book covers not only the 'inorganic' chemistry of the elements, but also analytical, theoretical, industrial, organometallic, bio-inorganic and other cognate areas of chemistry. The authors have broken with recent tradition in the teaching of their subject and adopted a new and highly successful approach based on descriptive chemistry. The chemistry of the elements is still discussed within the context of an underlying theoretical framework, giving cohesion and structure to the text, but at all times the chemical facts are emphasized. Students are invited to enter the exciting world of chemical phenomena with a sound knowledge and understanding of the subject, to approach experimentation with an open mind, and to assess observations reliably. This is a book that students will not only value during their formal education, but will keep and refer to throughout their careers as chemists. Completely revised and updated Unique approach to the subject More comprehensive than competing titles

**What Is a Chemical Element?** Eric Scerri, Elena Ghibaudi, 2020 This book offers a comprehensive overview of an important notion to the field of chemistry: the chemical element--

Chemical Elements in the Environment Clemens Reimann, Patrice de Caritat, 2012-12-06 How large is the natural variation in concentration of the various elements in different media? How do the oft-cited World average concentrations in different media compare with actual analytical data? How low a detection limit do I need to attain if I want to analyse for an element in soils, sediments, water or plants? All these questions and many more can be answered by using this unique reference book. It collates data on the most important properties and uses of all naturally occurring chemical elements. It combines these with data obtained from actual analyses of different sample media (soil, stream sediment, stream water, ground water, plants, human body fluids). This combination of facts and actual data makes this book suitable for learning and teaching applied geochemistry as well.

**Exploring Chemical Elements and Their Compounds** David L. Heiserman, 1992 Finding a book on the chemical elements that is neither an advanced, graduate-level text nor a simplistic overview for children is virtually impossible. Now, with *Exploring Chemical Elements and Their Compounds*, David L. Heiserman provides the perfect guide for anyone who needs a good solid introduction to all of the 107 chemical elements.

*Nature's Building Blocks* John Emsley, 2003 A readable, informative, fascinating entry on each one of the 100-odd chemical elements, arranged alphabetically from actinium to zirconium. Each entry comprises an explanation of where the element's name comes from, followed by Body element (the role it plays in living things), Element of history (how and when it was discovered), Economic element (what it is used for), Environmental element (where it occurs, how much), Chemical element (facts, figures and narrative), and Element of surprise (an amazing, little-known fact about it). A wonderful 'dipping into' source for the family reference shelf and for students.

**Encyclopedia of the Elements** Per Enghag, 2008-01-08 Famous for its history of numerous element discoverers, Sweden is the

origin of this comprehensive encyclopedia of the elements. It provides both an important database for professionals as well as detailed reading ranging from historical facts, discoverers' portraits, colour plates of mineral types, natural occurrences, and industrial figures to winning and refining processes, biological roles and applications in modern chemistry, engineering and industry. Elemental data is presented in fact tables which include numerous physical and thermodynamic properties, isotope lists, radiation absorption characteristics, NMR parameters, and others. Further pertinent data is supplied in additional tables throughout the text. Published in Swedish in three volumes from 1998 to 2000, the contents have been revised and expanded by the author for this English edition.

The Chemistry of Superheavy Elements Matthias Schädel, 2007-05-08 This book is the first to treat the chemistry of superheavy elements, including important related nuclear aspects, as a self contained topic. It is written for those – students and novices -- who begin to work and those who are working in this fascinating and challenging field of the heaviest and superheavy elements, for their lecturers, their advisers and for the practicing scientists in the field – chemists and physicists - as the most complete source of reference about our today's knowledge of the chemistry of transactinides and superheavy elements. However, besides a number of very detailed discussions for the experts this book shall also provide interesting and easy to read material for teachers who are interested in this subject, for those chemists and physicists who are not experts in the field and for our interested fellow scientists in adjacent fields. Special emphasis is laid on an extensive coverage of the original literature in the reference part of each of the eight chapters to facilitate further and deeper studies of specific aspects. The index for each chapter should provide help to easily find a desired topic and to use this book as a convenient source to get fast access to a desired topic. Superheavy elements – chemical elements which are much heavier than those which we know of from our daily life – are a persistent dream in human minds and the kernel of science fiction literature for about a century.

**The Elements**, 1991 Reference book presenting facts about the chemical elements, arranged in alphabetical order.

**On the Classification and the Atomic Weights of the So-called Chemical Elements, with Reference to Stas'**

**Determinations** Gustav Dethlef Hinrichs, 1870

*Exploring the Elements* Isabel Thomas, 2020 Science meets design in this comprehensive introduction to the chemical elements that make up our universe. This artful and accessible guide to the periodic table, the ultimate reference tool for scientists worldwide, names all 118 chemical elements and helps young readers understand the remarkable ways we have learned to use them. Graphically stunning layouts feature each element's letter symbol and atomic number, exploring its attributes, characteristics, uses, and interesting stories behind its discovery. Complete with a comprehensive introduction, conclusion, and glossary, this is the perfect introduction to chemistry for inquisitive minds.

What Is A Chemical Element? Eric Scerri, Elena Ghibaudi, 2020-05-11 The concept of a chemical element is foundational within the field of chemistry, but there is wide disagreement over its definition. Even the International Union for Pure and Applied Chemistry (IUPAC) claims two distinct definitions: a species of atoms versus one which identifies chemical elements with the simple substances bearing their names. The double definition of elements proposed by the International Union for Pure and Applied Chemistry contrasts an abstract meaning and an operational one. Nevertheless, the philosophical aspects of this notion are not fully captured by the IUPAC definitions, despite the fact that they were crucial for the construction of the Periodic Table. Although rich scientific literature on the element and the periodic table exists as well as a recent growth in the philosophy of chemistry, scholars are still searching for a

definitive answer to this important question: What is an element? Eric Scerri and Elena Ghibaudi have teamed up to assemble a group of scholars to provide readers an overview of the current state of the debate on chemical elements from epistemological, historical, and educational perspectives. *What Is A Chemical Element?* fills a gap for the benefit of the whole chemistry community-experimental researchers, philosophers, chemistry educators, and anyone looking to learn more about the elements of the periodic table.

**Reference Data on Atoms, Molecules, and Ions** A.A. Radzig, B.M. Smirnov, 2012-12-06 This reference book contains information about the structure and properties of atomic and molecular particles, as well as some of the nuclear parameters. It includes data which can be of use when studying atomic and molecular processes in the physics of gases, chemistry of gases and gas optics, in plasma physics and plasma chemistry, in physical chemistry and radiation chemistry, in geophysics, astrophysics, solid-state physics and a variety of cross-disciplinary fields of science and technology. Our aim was to collect carefully selected and estimated numerical values for a wide circle of microscopic parameters in a relatively not thick book. These values are of constant use in the work of practical investigators. In essence, the book represents a substantially revised and extended edition of our reference book published in Russian in 1980. Two main reasons made it necessary to rework the material. On the one hand, a great deal of new high-quality data has appeared in the past few years and furthermore we have enlisted many sources of information previously inaccessible to us. On the other hand, we have tried to insert extensive information on new, rapidly progressing branches of physical research, such as multiply charged ions, Rydberg atoms, van der Waals and excimer molecules, complex ions, etc. All this brings us to the very edge of studies being carried out in the field.

**Periodic Table Advanced** BarCharts, Inc, Mark Jackson, 2014-12-31 The ultimate reference tool and lab partner for any student of science, durably laminated and designed to fit as much info as possible in this handy 6-page format. Separate property tables are broken out for the ease of locating trends while studying and working while other pages offer essential notes about the table's organization and history. Consistently, a best seller since its first creation.

*30-Second Elements* Eric Scerri, 2013-08-01 When was radium discovered? Who are Dmitri Mendeleev and Glenn T. Seaborg? Who discovered uranium's radioactivity? Which element is useful for dating the age of Earth? And why doesn't gold have a scientific name? *30-Second Elements* presents you with the very foundations of chemical knowledge, explaining concisely the 50 most significant chemical elements. This book uses helpful glossaries and tables to fast track your knowledge of the other 68 elements and the relationships between all of them.

**Periodic Table with Nuclides and Reference Data** K. Yoshihara, H. Kudo, T. Sekine, 2012-05-19 There are many kinds of nuclear data books; however some are too much specialized, while others have an arrangement of information which is inconvenient for students to use. With this book, we want to amend these situations. Handbooks of natural sciences must be exact and fair in their presentation of materials and they must be logical and convenient to use. If the users can develop new ideas or gain new insights from the books, they can be judged as valuable. The role of handbooks is not only to give a systematic representation of past knowledge, but also to serve as a basis for intellectual activity leading to future development. The purpose of this data book arises from the points described above. The chart of the nuclides which is frequently consulted by radioisotope users is not always convenient. By comparison, our Periodic Table with Nuclides has been devised with this in mind. It has been our experience that properties of a desired nuclide could be found in a much shorter time in the Periodic Table with Nuclides than in other nuclide charts. Additionally,

by placing the ~-stability line within the nuclides in the table, the users may derive unambiguous ideas on the stability of the nuclides and the paths related to the creation of stable elements in the universe.

The Chemical Elements in Nature Frank Henry Day, 1965

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## Chemical Elements Reference Introduction

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