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identify the limiting reactant s and excess reactant s the limiting reactant is o₂ since it would yield the least amount of product 18 mol br₂ the excess reactant is c₂h₃br₃ since its

complete reaction would have yielded up to 54 mol br₂ think about your result

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web if you start with 14 82 g of Ca(OH)_2 c a o h 2 and 16 35 g of H_2SO_4 h 2 s o 4 a determine the limiting reagent b determine the number of

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