

# **Periodic Table**

**Paul Parsons**

*The Periodic Table Book* DK,2017-03-30 The Periodic Table Book is the perfect visual guide to the chemical elements that make up our world. This eye-catching encyclopedia takes children on a visual tour of the 118 chemical elements of the periodic table, from argon to zinc. It explores the naturally occurring elements, as well as the man-made ones, and explains their properties and atomic structures. Using more than 1,000 full-colour photographs, The Periodic Table Book shows the many natural forms of each element, as well as a wide range of both everyday and unexpected objects in which it is found, making each element relevant for the child's world.

*Periodic Table, The: Past, Present, And Future* Geoffrey Rayner-canham,2020-08-04 'This is an absolutely wonderful book that is full of gems about the elements and the periodic table ... All in all, the book is highly recommended to philosophers of chemistry. As philosophers we have a natural tendency to concentrate on generalities and not to get too involved in the specifics and the details. Above all else, this new book reminds us that such an approach needs to be tempered by a detailed knowledge of the exceptions and features that go against the simplified generalities which we so cherish.' [Read Full Review]Eric ScerriFoundations of Chemistry'Many questions are dealt with in a clearly written way in this stimulating and innovative book. The reader will quickly become interested in the subject and will be taken on tour through this Periodic Table in a very readable way, both for students and teachers ... The number of illustrations is good, and clear. This book is indeed unique and quite thought-provoking ... This book is highly recommended for students, teachers, researchers and not only chemists! Geologists, biochemists and also physicists will find it very interesting to read.' [Read Full Review]Chemistry InternationalThat fossilized chart on every classroom wall — isn't that The Periodic Table? Isn't that what Mendeléev devised about a century ago? No and No. There are many ways of organizing the chemical elements, some of which are thought-provoking, and which reveal philosophical challenges. Where does hydrogen 'belong'? Can an element occupy more than one location on the chart? Which are the Group 3 elements? Is aluminum in the wrong place? Why is silver(I) like thallium(I)? Why is vanadium like molybdenum? Why does gold form an auride ion like a halide ion? Does an atom 'know' if it is a non-metal or metal? Which elements are the 'metalloids'? Which are the triels? So many questions! In this stimulating and innovative book, the Reader will be taken on a voyage from the past to the present to the future of the Periodic Table. This book is unique. This book is readable. This book is thought-provoking. It is a multi-dimensional examination of patterns and trends among the chemical elements. Every reader will discover something about the chemical elements which will provoke thought and a new appreciation as to how the elements relate together.

**The Periodic Table** Primo Levi,1996-10-01 The Periodic Table is largely a memoir of the years before and after Primo Levi's transportation from his native Italy to Auschwitz as an anti-Fascist partisan and a Jew. It recounts, in clear, precise, unfailingly beautiful prose, the story of the Piedmontese Jewish community from which Levi came, of his years as a student and young chemist at the inception of the Second World War, and of his investigations into the nature of the material world. As such, it provides crucial links and backgrounds, both personal and intellectual, in the tremendous project of remembrance that is Levi's gift to posterity. But far from being a prologue to his experience of the Holocaust, Levi's masterpiece represents his most impassioned response to the events that engulfed him. The Periodic Table celebrates the pleasures of love and friendship and the search for meaning, and stands as a monument to those things in us that are capable of resisting and enduring in the face of tyranny.

*The Periodic Table* Tom Jackson,2020-02-04 Which is the densest element? Which has the largest atoms? And why are some elements radioactive? From the little-known uses of gold in medicine to the development of the hydrogen bomb, this is a fresh new look at the Periodic Table. Combining cutting edge science with fascinating facts and stunning infographics, this book looks at the extraordinary stories of discovery, amazing properties and surprising uses of each elements, whether solid, liquid or gas - naturally occurring, synthesised or theoretical! From hydrogen to oganesson, this is a fact-filled visual guide to each element,each accompanied by technical date

(category, atomic number, weight, boiling point) as well as fun facts and stories about their discovery and surprising uses.

**The Periodic Table** Sharon Katz Cooper, 2007 An introduction to chemistry's periodic table of elements.

**The Periodic Table** Eric Scerri, 2019-10-21 The periodic table of elements is among the most recognizable image in science. It lies at the core of chemistry and embodies the most fundamental principles of science. In this new edition, Eric Scerri offers readers a complete and updated history and philosophy of the periodic table. Written in a lively style to appeal to experts and interested laypersons alike, *The Periodic Table: Its Story and Its Significance* begins with an overview of the importance of the periodic table and the manner in which the term element has been interpreted by chemists and philosophers across time. The book traces the evolution and development of the periodic table from its early beginnings with the work of the precursors like De Chancourtois, Newlands and Meyer to Mendeleev's 1869 first published table and beyond. Several chapters are devoted to developments in 20th century physics, especially quantum mechanics and the extent to which they explain the periodic table in a more fundamental way. Other chapters examine the formation of the elements, nuclear structure, the discovery of the last seven infra-uranium elements, and the synthesis of trans-uranium elements. Finally, the book considers the many different ways of representing the periodic system and the quest for an optimal arrangement.

**The Periodic Table** Paul Parsons, 2014-03-11 As one of the most recognizable images in science, the periodic table is ingrained in our culture. First drawn up in 1869 by Dmitri Mendeleev, its 118 elements make up not only everything on our planet but also everything in the entire universe. The Periodic Table looks at the fascinating story and surprising uses of each of those elements, whether solid, liquid or gas. From the little-known uses of gold in medicine to the development of the hydrogen bomb, each entry is accompanied by technical data (category, atomic number, weight, boiling point) presented in easy-to-read headers, and a colour coding system that helps the reader to navigate through the different groups of elements. A remarkable display of thought-provoking science and beautiful photography, this guide will allow the reader to discover the world afresh.

**The Periodic Table** Eric R. Scerri, 2019 The periodic table of elements, first encountered by many of us at school, provides an arrangement of the chemical elements, ordered by their atomic number, electron configuration, and recurring chemical properties, and divided into periodic trends. In this Very Short Introduction Eric R. Scerri looks at the trends in properties of elements that led to the construction of the table, and shows how the deeper meaning of the table's structure gradually became apparent with the development of atomic theory and, in particular, quantum mechanics, which underlies the behaviour of all of the elements and their compounds. This new edition, publishing in the International Year of the Periodic Table, celebrates the completion of the seventh period of the table, with the ratification and naming of elements 113, 115, 117, and 118 as nihonium, moscovium, tennessine, and oganesson. Eric R. Scerri also incorporates new material on recent advances in our understanding of the origin of the elements, as well as developments concerning group three of the periodic table. **ABOUT THE SERIES:** The Very Short Introductions series from Oxford University Press contains hundreds of titles in almost every subject area. These pocket-sized books are the perfect way to get ahead in a new subject quickly. Our expert authors combine facts, analysis, perspective, new ideas, and enthusiasm to make interesting and challenging topics highly readable.

**A Brief History of the Periodic Table** Harish Chandra Rai, 2019-12-11 The story of Dmitri Ivanovich Mendeleev and his brain child "Periodic Table of Chemical Elements", with all its impact and influences, would fit better within the walls of a library than between the covers of a single book of nearly 100 pages. The present book "A Brief History of the Periodic Table" would attract experts and curious laymen alike due to its lively style of narration. The book contains eight chapters.

**Mystery of the Periodic Table** Benjamin D Wiker, 2003-04-18 Leads the reader on a delightful and absorbing journey through the ages, on the trail of the elements of the Periodic Table as we

know them today. He introduces the young reader to people like Von Helmont, Boyle, Stahl, Priestly, Cavendish, Lavoisier, and many others, all incredibly diverse in personality and approach, who have laid the groundwork for a search that is still unfolding to this day. The first part of Wiker's witty and solidly instructive presentation is most suitable to middle school age, while the later chapters are designed for ages 12-13 and up, with a final chapter somewhat more advanced. Illustrated by Jeanne Bendick and Ted Schluenderfritz.

**The Periodic Table of the Elements** Richard J. Puddephatt, Patrick Kevin Monaghan, 1994

**The Periodic Table in Minutes** Dan Green, 2016-09-06 An icon of science, the Periodic Table defines the fundamental chemistry of everything in the universe. In this compact yet comprehensive guide, Dan Green outlines the history, development and workings of the table, shows how its design reflects and illuminates the organisation of all matter, and even explains what it has to tell us about the chemistry of distant stars and of our own bodies. Contents include an individual entry for every known element? detailing properties, uses and key data, and sections on the patterns and groups of the famous table, as well as explanations of basic chemistry concepts such as elements and compounds, atomic structure, chemical bonds, reactions and radioactivity, amongst many others.

*Periodic Table & Periodic Properties* M. Satake, 2010 The book is primarily meant for undergraduate students of chemistry. General reader who is interested in chemistry of elements and their behaviour will find it equally interesting and easy to understand.

*The Periodic Table* Eric R. Scerri, 2006-10-12 The periodic table is one of the most potent icons in science. It lies at the core of chemistry and embodies the most fundamental principles of the field. The one definitive text on the development of the periodic table by van Spronsen (1969), has been out of print for a considerable time. The present book provides a successor to van Spronsen, but goes further in giving an evaluation of the extent to which modern physics has, or has not, explained the periodic system. The book is written in a lively style to appeal to experts and interested laypersons alike. The Periodic Table begins with an overview of the importance of the periodic table and of the elements and it examines the manner in which the term 'element' has been interpreted by chemists and philosophers. The book then turns to a systematic account of the early developments that led to the classification of the elements including the work of Lavoisier, Boyle and Dalton and Cannizzaro. The precursors to the periodic system, like D?bereiner and Gmelin, are discussed. In chapter 3 the discovery of the periodic system by six independent scientists is examined in detail. Two chapters are devoted to the discoveries of Mendeleev, the leading discoverer, including his predictions of new elements and his accommodation of already existing elements. Chapters 6 and 7 consider the impact of physics including the discoveries of radioactivity and isotopy and successive theories of the electron including Bohr's quantum theoretical approach. Chapter 8 discusses the response to the new physical theories by chemists such as Lewis and Bury who were able to draw on detailed chemical knowledge to correct some of the early electronic configurations published by Bohr and others. Chapter 9 provides a critical analysis of the extent to which modern quantum mechanics is, or is not, able to explain the periodic system from first principles. Finally, chapter 10 considers the way that the elements evolved following the Big Bang and in the interior of stars. The book closes with an examination of further chemical aspects including lesser known trends within the periodic system such as the knight's move relationship and secondary periodicity, as well as attempts to explain such trends.

**The Periodic Table** Mary Griffin, 2018-12-15 The periodic table is the way scientists have organized the known elements, but up-and-coming scientists can learn much from the periodic table as well. Your readers will learn how to read the periodic table, including what an element's atomic number means and what the rows and columns signify. They'll gather facts about common and interesting elements and even discover how they can predict how elements might chemically react from their position in the table. Fact boxes and images add additional information to the accessible text.

*Elementary* James M. Russell, 2019-06-13 The periodic table, created in the early 1860s by Russian chemist Dmitri Mendeleev, marked one of the most extraordinary advances in modern

chemistry. This basic visual aid helped scientists to gain a deeper understanding of what chemical elements really were: and, astonishingly, it also correctly predicted the properties of elements that hadn't been discovered at the time. Here, in the authoritative Elementary, James Russell uses his lively, accessible and engaging narrative to tell the story behind all the elements we now know about. From learning about the creation of the first three elements, hydrogen, lithium and helium, in the big bang, through to oxygen and carbon, which sustain life on earth - along with the many weird and wonderful uses of elements as varied as fluorine, arsenic, krypton and einsteinium - even the most unscientifically minded will be enthralled by this fascinating subject. Russell compellingly details these most basic building blocks of the universe, and the people who identified, isolated and even created them.

**Animated Science: Periodic Table** John Farndon, 2021-10-05 The elements of the periodic table come alive in the first book in a stellar nonfiction comic series illustrated by Shiho Pate! From oxygen to hydrogen, carbon to plutonium, Animated Science: Periodic Table makes chemistry come alive! In this book you'll meet the building blocks of you, the world, and the universe and see how they come together to make everything you see, do, and use every day. With a narrative nonfiction text, kid-friendly information, and Shiho Pate's engaging illustrations, Animated Science: Periodic Table is a perfect introduction and ready reference, appealing and laugh-out-loud funny. Easily accessible for readers just learning the elements, with more interesting facts and details for older kids honing their knowledge. Great for all ages!

*Basher Science: The Complete Periodic Table* Adrian Dingle, Simon Basher, Dan Green, 2017-01-31 Do you confuse boron with barium or chlorine with fluorine? Fear not! Basher Science has come to the rescue by mixing science and art to create a unique periodic table. From unassuming oxygen to devious manganese, the incredible elements show you the periodic table as you've never seen it before. Basher Science: The Periodic Table gives a face, voice and personality to the chemical elements, making learning chemistry easy and a whole lot more fun. This new expanded edition reflects the latest discoveries, and now each of the 115 elements has not just a picture but an information-packed page all to itself. Basher's highly original books make difficult concepts tangible, understandable and even lovable. With his stylish, contemporary characters he communicates science brilliantly.

**The Lost Elements** Marco Fontani, Mariagrazia Costa, Mary Virginia Orna, 2015 In the mid-nineteenth century, chemists came to the conclusion that elements should be organized by their atomic weights. However, the atomic weights of various elements were calculated erroneously, and chemists also observed some anomalies in the properties of other elements. Over time, it became clear that the periodic table as currently comprised contained gaps, missing elements that had yet to be discovered. A rush to discover these missing pieces followed, and a seemingly endless amount of elemental discoveries were proclaimed and brought into laboratories. It wasn't until the discovery of the atomic number in 1913 that chemists were able to begin making sense of what did and what did not belong on the periodic table, but even then, the discovery of radioactivity convoluted the definition of an element further. Throughout its formation, the periodic table has seen false entries, good-faith errors, retractions, and dead ends; in fact, there have been more elemental discoveries that have proven false than there are current elements on the table. *The Lost Elements: The Shadow Side of Discovery* collects the most notable of these instances, stretching from the nineteenth century to the present. The book tells the story of how scientists have come to understand elements, by discussing the failed theories and false discoveries that shaped the path of scientific progress. Chapters range from early chemists' stubborn refusal to disregard alchemy as legitimate practice, to the effects of the atomic number on discovery, to the switch in influence from chemists to physicists, as elements began to be artificially created in the twentieth century. Along the way, Fontani, Costa, and Orna introduce us to the key figures in the development of the periodic table as we know it. And we learn, in the end, that this development was shaped by errors and gaffs as much as by correct assumptions and scientific conclusions.

**Essential Elements** Matt Tweed, 2003-04-01 A short, illustrated introduction to the tiny

building blocks of our universe including atoms, quarks, and the periodic table. Illustrations. 10,000 print.

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