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that between a  
metal and an  
acid when a  
metal reacts  
with an acid  
the reaction  
produces a salt  
and hydrogen  
gas is released  
the salt  
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depends upon  
the metal and  
the acid if  
magnesium  
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web write the  
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equation for  
the combustion  
of  $\text{C}_2\text{H}_2$  in  
oxygen  $2\text{C}_2\text{H}_2 + 5\text{O}_2 \rightarrow 4\text{CO}_2 + 2\text{H}_2\text{O}$   
if 1.0 mol of  
 $\text{C}_2\text{H}_2$  is burned  
how many moles  
of  $\text{CO}_2$  are  
formed 2.5 mol  
if 1.0 mol of  
 $\text{C}_2\text{H}_2$  is burned  
how many moles  
of oxygen gas  
are consumed 9  
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a chemical

reaction  
equation gives  
the reactants  
and products  
and a balanced  
chemical  
reaction  
equation shows  
the mole  
relationships  
of reactants  
and products  
often the  
amount of  
energy involved  
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are used to describe chemical reactions look at the word equations below in each case complete the word equation by adding the name of the missing substance and explain your answers nitric acid potassium hydroxide water zinc zinc nitrate copper

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web derive chemical equations from narrative descriptions of chemical reactions write and balance chemical

equations in molecular total ionic and net ionic formats an earlier chapter of this text introduced the use of element symbols to represent individual atoms

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write a balanced molecular equation describing each of the following chemical reactions solid calcium carbonate is heated and decomposes to solid calcium

oxide and carbon dioxide gas gaseous butane c 4 h 10 reacts with diatomic oxygen gas to yield gaseous carbon dioxide and water vapor

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web chemical reactions answer key 1 watch the video and then complete the chart teacher notes on next page colors a red b blue c green d yellow 2 use colored pencils to circle the common atoms or compounds on each side of the equations to help you determine the

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